

DESCRIPTION OF MATTER (mass and occupies space) **Chp 1.1 & 1.2**

PURE (always homogeneous)	MIXTURE (impure)									
<p>Elements – all atoms are the same</p> <p>Represented by symbols of one element</p> <p>Compounds – made from two or more elements</p> <p>Represented by symbols of two or more elements</p>	<p>homogeneous (uniform throughout – composition/properties)</p> <table border="1" data-bbox="597 457 925 737"> <tr> <td data-bbox="597 457 763 632">brass</td> <td data-bbox="766 457 925 632">sugar dissolved in water</td> </tr> <tr> <td data-bbox="597 636 763 737">air</td> <td data-bbox="766 636 925 737">tap water</td> </tr> </table>	brass	sugar dissolved in water	air	tap water	<p>heterogeneous (not uniform composition/properties)</p> <table border="1" data-bbox="954 457 1291 768"> <tr> <td data-bbox="954 457 1120 596">granite</td> <td data-bbox="1123 457 1291 596">fog (air water)</td> </tr> <tr> <td data-bbox="954 600 1120 768">dirt</td> <td data-bbox="1123 600 1291 768">fresh squeezed lemonade (pulp in it)</td> </tr> </table>	granite	fog (air water)	dirt	fresh squeezed lemonade (pulp in it)
brass	sugar dissolved in water									
air	tap water									
granite	fog (air water)									
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STATES OF MATTER - gas, liquid, solid **Chp 1.4**

TRANSFORMATION of matter **Chp 1.6**

<p><u>Physical – phase change</u></p> <p>ice melting</p> <p>grinding coffee beans</p>	<p><u>Chemical- transformation to new substance(s)</u></p> <p>Hydrogen reacting with oxygen to produce water</p> <p>Hydrogen peroxide decomposing to water and oxygen</p>
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PROPERTIES of matter **Chp 1.6**

<p><u>Physical (detected using the senses):</u></p> <p>Boiling point</p> <p>Melting point</p> <p>Shape</p> <p>Color</p>	<p><u>Chemical (stability):</u></p> <p>Reactivity with oxygen</p> <p>Stability in air</p> <p>Heat stability</p>
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## HOW ARE PROPERTIES AND TRANSFORMATION OF MATTER STUDIED?

- HYPOTHESIS – TENTATIVE EXPLANATION OF AN OBSERVATION THAT CAN BE VERIFIED BY EXPERIMENT. *Experiments must be conducted under carefully controlled conditions so that they can be reproduced over and over. This separates science from pseudo science.*
- THEORY – AN EXPLANATION OF AN OBSERVATION THAT IS SUPPORTED BY EXPERIMENTAL DATA AND WHICH IS USED TO PREDICT OTHER OBSERVATIONS. PHYSICAL OR MATHEMATICAL MODELS ARE FREQUENTLY DERIVED FROM A THEORY. *At any time, a new experiment may be done that invalidates the theory.*
- LAW – A STATEMENT OF OBSERVATION OF A NATURAL PHENOMENA (Law of Conservation of Mass - “The total mass of reactants is equal to the total mass of products in a chemical reaction.”) *Laws do not ever change because they are based on observations of natural phenomena.*